

# Novel Iron Doped Materials for Mid-Infrared Laser Applications

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## Introduction

Recent progress in technology of Cr:ZnSe and Fe:ZnSe gain media enabled efficient lasers operating over 1.8-3.3 and 3.6-5.2  $\mu\text{m}$ . Nevertheless, there is strong demand for effective gain materials for the 3.0-3.9  $\mu\text{m}$  spectral range not nicely covered by Cr:ZnSe and Fe:ZnSe amplification bands. Therefore, a search of new perspective materials for laser applications in this spectral range is in a great demand. The  $^5D$  is ground states of the  $\text{Fe}^{2+}$  ions which is split into the doublet  $^5E$  and triplet  $^5T_2$  in the crystal field. The order and energy difference between the  $^5E$  state and the  $^5T_2$  states depends on ionic valence state, bond lengths and crystal field symmetry.

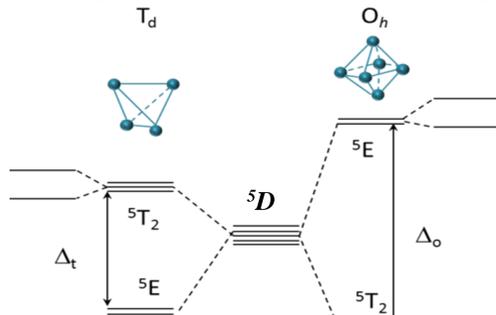


Figure 1. Energy Levels of  $\text{Fe}^{2+}$  ions in tetrahedral and octahedral sites

The doublet  $^5E$  is the ground state of the  $\text{Fe}^{2+}$  ions in the tetrahedral sites, while the triplet  $^5T_2$  is the ground state in the octahedral sites. In addition, the crystal field theory predicts approximately twice smaller energy splitting in tetrahedral site in comparison with octahedral site [1,2]. Therefore, crystal with tetrahedral coordination of iron ions are more promising for mid-IR applications.

In this research, we characterized Fe:MgAl<sub>2</sub>O<sub>4</sub> (spinel), Fe:ZnAl<sub>2</sub>O<sub>4</sub> (gahnite) and Fe:InP crystals as promising laser materials for this mid-IR spectral range.

## Samples Fabrications

MgAl<sub>2</sub>O<sub>4</sub> and ZnAl<sub>2</sub>O<sub>4</sub> crystals are the most representative crystals from oxide spinel group with chemical formula AB<sub>2</sub>O<sub>4</sub> where A and B metal ions have valence +2 and +3 correspondingly. Oxygen ions form face-centered cubic structure with 32 octahedral ( $O_h$ ) and 64 tetrahedral ( $T_d$ ) sites inside unit cell. In the normal spinel, 1/8 of tetrahedral sites are occupied by A<sup>2+</sup> ions and half of the  $O_h$  occupied by B<sup>3+</sup> ions.

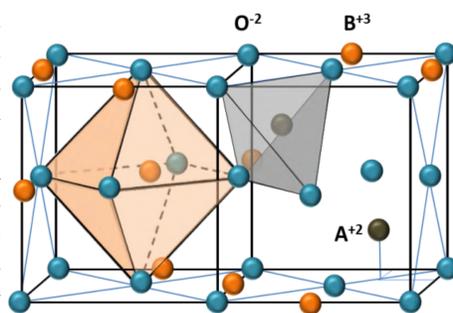
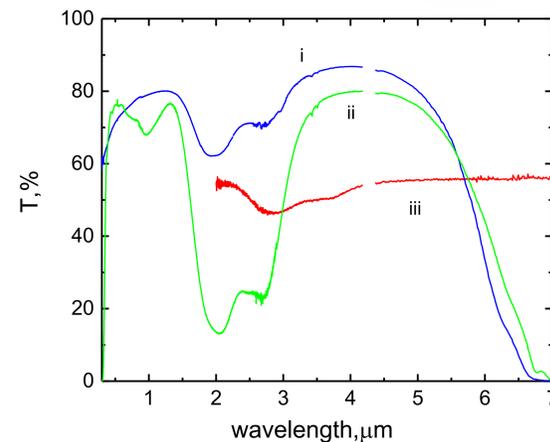


Figure 2. Crystal structure of spinel crystals

In our study we used Fe:MgAl<sub>2</sub>O<sub>4</sub> single crystal doped by thermal diffusion methods and Fe:ZnAl<sub>2</sub>O<sub>4</sub> ceramic samples doped during fabrication process.

Indium phosphide (InP) crystals have zincblende crystal structure where each ion is located in a tetrahedral site consisting of four ions of the opposite type [1]. In our study we used Fe:InP single crystal doped during crystal growth process with nominal concentration of iron  $10^{18} \text{ cm}^{-3}$ .

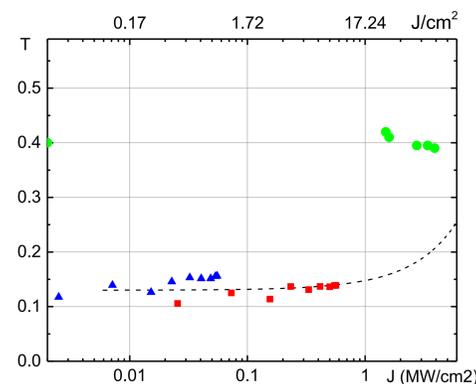
## Absorption measurements



- Spectroscopy Analysis:**
- Visible and IR absorption spectra were measured at room temperature using Shimadzu UV-VIS-NIR-3101PC and Prestige-21 (Shimadzu) FTIR spectrometer.

Figure 3. Transmission spectra of Fe:MgAl<sub>2</sub>O<sub>4</sub> (curve i); Fe:ZnAl<sub>2</sub>O<sub>4</sub> (curve ii); and Fe:InP (curve iii) samples.

## Saturation of Absorption



- Experimental setup:**
- Saturation of the absorption in Fe:ZnAl<sub>2</sub>O<sub>4</sub> samples was studied under excitation by radiations of Q-switched Ho:YAG @ 2.09  $\mu\text{m}$  and Er:YAG @ 2.9  $\mu\text{m}$  lasers

Figure 4. Transmission spectra of Fe:ZnAl<sub>2</sub>O<sub>4</sub> sample vs pump energy density at 2.09  $\mu\text{m}$  wavelength (circles) and 2.94  $\mu\text{m}$  wavelength (triangles and squares); dashed line shows theoretical dependence with upper level life time 50 ps.

## Photoluminescence measurements

- Experimental setup:**
- The non-selective mid-IR photoluminescence of  $\text{Fe}^{2+}$  ions was measured using HgCdTe detector with temporal resolution  $\sim 80 \text{ ns}$ .
  - The long-wavelength pass optical filter with cut-off wavelength at 3.5  $\mu\text{m}$  was installed before HgCdTe detector.
  - The Q-switched Er:YAG @ 2.94  $\mu\text{m}$  and Ho:YAG @ 2.09  $\mu\text{m}$  laser were used for direct excitation of the  $^5E \rightarrow ^5T_2$  transition of  $\text{Fe}^{2+}$  ions in the tetrahedral sites.
  - A Q-switched Nd:YAG @ 1.064  $\mu\text{m}$  laser was used to study excitation of  $\text{Fe}^{2+}$  ions in the InP crystal via photoionization transition

## Discussion

- In addition to broad absorption band of  $\text{Fe}^{2+}$  ions in tetrahedral sites between 2-3  $\mu\text{m}$ , the absorption spectra of Fe:ZnAl<sub>2</sub>O<sub>4</sub> ceramic samples reveal  $\text{Fe}^{2+}$  ions in octahedral sites with maximum of the absorption band around 1  $\mu\text{m}$ .
- The thermal diffusion of iron in MgAl<sub>2</sub>O<sub>4</sub> single crystal could be used for fabrication of Fe:MgAl<sub>2</sub>O<sub>4</sub> samples with ion concentration  $\sim 5 \times 10^{19} \text{ cm}^{-3}$ .
- The radiative life time for  $^5E \rightarrow ^5T_2$  transition of  $\text{Fe}^{2+}$  ions in tetrahedral sites in the MgAl<sub>2</sub>O<sub>4</sub> crystal was estimated using following equation:

$$\tau_{rad} = \frac{1}{8\pi c n^2} \left[ \frac{g_1}{g_2} \int \tilde{\nu}^2 \sigma_{abs} d\tilde{\nu} \right]^{-1}$$

, and was calculated to be 35  $\mu\text{s}$ .

- Absence of the absorption saturation (up to pump power density  $\sim 1 \text{ MW/cm}^2$ ) and absence of the mid-IR photoluminescence in Fe:ZnAl<sub>2</sub>O<sub>4</sub> sample at room temperature, allow estimation of the nonradioactive decay time faster than 50 ps
- The photoluminescence signals in Fe:InP crystal were detected under direct excitation of  $\text{Fe}^{2+}$  ions by radiation at 2.94  $\mu\text{m}$  as well as under excitation via photo-ionization transition by 1.064  $\mu\text{m}$  radiation.

## Conclusions and Future Work

- Fe:InP has the potential to be an effective mid-IR laser material, but would require optimization of material doping technology to increase concentration of  $\text{Fe}^{2+}$  ions to  $\sim 10^{19} \text{ cm}^{-3}$ .
- Due to strong quenching of the mid-IR photoluminescence, Fe:ZnAl<sub>2</sub>O<sub>4</sub> and Fe:MgAl<sub>2</sub>O<sub>4</sub> crystals are not suitable for laser application at room temperature. The possibility of utilization these crystal should be tested at low temperature.

## References

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